

**DOSE RESPONSE STUDY OF N, N'-DIMETHYLUREA (DMU) ANTIDOTE TOTAL FOLIAR PROTEIN IN *Lycopersicon esculentum* AGAINST THE EFFECT OF O<sub>3</sub>****EKTA DUBEY<sup>a1</sup> AND P. S. DUBEY<sup>b</sup>**<sup>a</sup>Department of Botany, Govt. Arts and Science P.G. College, Vikram University, Ratlam, Madhya Pradesh, India<sup>b</sup>Former Professor and Head IEMPS (Institute of Environment Management and Plant Sciences, Ex- Chairman MPPCB (Madhya Pradesh Pollution Control Board Bhopal), Madhya Pradesh, India**ABSTRACT**

To overcome the problem of agriculture loss due to air pollution, an alternative method would involve the application of protective chemical components to plants in the field to prevent injury. The studies of the potential for protection of vegetation from the oxidative gases using chemical application have been performed over the last three to four decades. This would allow the plant to grow under completely natural conditions and to be exposed to fluctuating natural concentrations of ground level ozone. Such diverse groups of chemical compounds are known as antidotes. There will always be a need to evaluate the quantitative and qualitative loss in various crops and to mitigate it by inducing scavenging potential (tolerance) in the crop plants against air pollution stress through the application of certain plant protectants. The protein level in plant is more important due to its participation in enzyme structure that plays key role in various metabolic activities of plant growth, development and yield Protein synthesis a building up process in plants is also prone to O<sub>3</sub> toxicity. Considerable evidence indicates that the free radicals can inactivate proteins by modifying amino acid residues. Loss in Protein content was found in Tomato crop against the gaseous concentration simultaneously significant recovery appeared against O<sub>3</sub> gaseous exposure. This may be because the treatment adds to the internal/ inherent tolerance capacity to the crop plants, which may help in the reduction of losses. The present study, decrease in protein content in 51 days Tomato plant age was recorded. The loss in Tomato plant was 16.45 %. After 62 days Plant age N, N'-Dimethylurea (DMU) used during the study was found effective and could reduce the deleterious effect of ozone in Tomato crops. The recovery observed after loss in Tomato plant low dose and high dose of SE brings recovery of 49.24% and 52.84% respectively. N, N'-Dimethylurea (DMU). was found best for the amelioration of ozone toxicity.

**KEYWORDS :** Amelioration, Antidotes, Environmental Pollution, Ozone, *Lycopersicon esculentum*, N, N'-Dimethylurea (DMU)

Atmospheric pollution is a major problem taking all nations of the world. Rapid urban and industrial growth has resulted in vast quantities of potentially harmful waste products being released into the atmosphere. The consequence has been that air pollution has caused widespread damaged to vegetations, crops, wildlife, building materials and climate. Air pollutants interact in variety of way with all living things. Since plants are dominant life forms in number and size over most of the earth surface, they interact with gaseous and particulate pollutants. Some plants absorb the pollutant with no concomitant effect on their growth and functioning while some were considered to be relatively more sensitive. Increasingly physiological, structural and biochemical effects or injuries are being used in the pollution studies. Depending upon the concentration and duration of gaseous exposure, plant response is arbitrarily classified in to three categories, acute injury, chronic injury and physiological injury. Injuries of plant finally result from the exhaustion of its repairing and compensating potential, which leads to faster alteration of biochemical pathways.

Ozone is a strong, gaseous oxidizing agent, which reacts rapidly with other atmospheric constituents. Photochemical oxidants result from atmospheric interactions of NO<sub>2</sub> and sunlight. Ozone is relatively unstable and releases its third oxygen atom readily so that it oxidizes or burns things more readily and at lower concentration than does normal oxygen. Chemically ozone is very active; it has a short average lifetime in the air because of the effect of sunlight on normal oxygen. Ozone forms a natural layer high in the atmosphere (Stratosphere) this ozone layer protects us from UV. The formation of ozone is the result of complex non-linear physical and chemical processes in the atmosphere including two classes of precursor pollutants NO<sub>2</sub> (Nitrogen dioxides) and VOC (Volatile Organic Compounds) Thus, in the lower atmosphere, ozone is a secondary pollutant formed from gaseous precursors. The effect of ozone on plants can be subtle. At very low concentration ozone can reduce growth rate while not producing visible injury. At high concentration ozone kills leaf tissue, killing entire leaves and if the pollutant level remains high, killing whole plant.

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Experiments with ozone protectant chemicals have also indicated that ozone can cause significant effects on the yield of *Lycopersicon esculentum* in and around New Delhi, on radish and turnip yields at a rural site in the Nile delta (Hassan et al., 1995), on yields of *Phaseolus vulgaris* in the Valley of Mexico (Ashmore and Marshall 1999; De Bauer 2003), and on soybean (*Glycine max*) in the Pakistan Punjab (Wahid et al. 2001). Recent data demonstrate that rural ozone levels are high enough to potentially affect yields of winter wheat (Chameides et al. 1999) and other crops (Zheng et al. 1998).

## MATERIALS AND METHODS

Experiments were conducted at the Institute of Environment Management and Plant Sciences, Vikram University Ujjain (M.P.). N, N'-Dimethylurea (DMU) was selected as Antidotes and *Lycopersicon esculentum* Mill (Tomato) (cv. Pusa samrat, Family Solanaceae) crop was selected to Experiments.

### O<sub>3</sub> (60 µg/m<sup>3</sup>) Gas treatment

#### Experimental Set

Ten healthy seeds of Tomato crop were sown in earthen pots separately containing 3 kg. Black Cotton Soil. After 10 days of germination and plant growth, thinning was carried out and 4-5 plantlets were allowed to grow further in each pot. Total 32 numbers of pots for Tomato crop was sown for spray treatment. After one month of normal growth plants were subjected to different combination and concentration of gaseous pollutant at the rate of 6hrs/ day. A control set was also run simultaneously. Each set was run with three replicates.

#### Mode of Application of Antidotes

Solution of concentration of 20 ppm and 100 ppm referred as low dose and high dose respectively was prepared of Sodium Erythroate antidotes (SE) with acetone and distilled water. 100 ml of each solution was sprayed with the calibrated spray gun on plants at the age of 51 and 83 days.

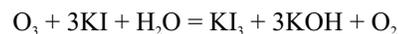
#### Instruments Employed in Study

Open Top Chamber (OTC) type of study has been carried out for treatment of O<sub>3</sub> of gaseous pollutant in this investigation. The crop plants were exposed to gaseous

pollutant in chamber. It was covered with transparent polythene. The top of the chamber was left open. All the crop plants to be exposed and considered to be control were placed in the chamber to obtain similar environmental condition with or without fumigation. The gas generated pass in the open top chamber through inner side Teflon tube in which holes of 1mm at 30 cm apart done.

#### Ozone (O<sub>3</sub>) Generation

Through UV- B exposure of air by Byers & Saltzman, 1958) - Micro amount of ozone liberate iodine when absorbed in a 1% solution of potassium iodide buffered at pH 6.80±2. The iodine is determined spectrophotometrically by measuring the absorption of triiodide ion at 352nm. The stoichiometry is approximated by the following reaction:



Ozone was generated by flowing the air through UV-Tubes at fixed rate of flow rate. The amount of gas generated was monitored by standard method Byers & Saltzman, 1958, with the help of Portable gas sampler by using the 1% potassium buffer solution and absorbance was read at 352nm at UV-VIS spectrophotometer. Concentration of ozone in the air was calculated with the help of a standard graph prepared by diluting a stock solution of 0.025 M iodine.

#### Estimation of Protein Content (Lowry et.al., 1951)

Protein content in the leaves was estimated by taking the 1mg sample, crushed with NaOH and centrifuge at 3000 rpm for 10 min. After this 0.2 ml supernatant was collected in centrifuge tube and distilled water and trichloro acetic acid were added and again centrifuge at same rpm. Now sediment was dissolved in 2% Na<sub>2</sub>CO<sub>3</sub> solution and folin reagent was added and blue colour appears and intensity was to be read at 750 nm and concentration was calculated with standard graph. Standard graph was prepared by dissolving crystalline bovine albumin.

## RESULTS AND DISCUSSION

Numerous techniques are available for identifying or measuring response of plants to oxidative stress such as physiological and biochemical changes. Many of these changes in stomatal conductance photosynthesis leaf

pigments, proteins, and lipids or other macromolecules and metabolite pool can be used as indicator for assessment of pollutant effects on vegetation and crops (Heath et al., 1997).

#### Amelioration Study of Foliar Protein Content

Proteins are synthesized in plants with the help of amino acids. Plant absorbs nitrogen in the form of NO<sub>3</sub>. This is subjected to reduction, so NO<sub>3</sub> is converted into NH<sub>3</sub>. This NH<sub>3</sub> reacts with carbonic acid to form amino acids. Finally, amino acids condense, to form proteins.

#### Plant Age 51 Days

The effect of oxidative pollutant on plants occurs when they enters the plant leaves disrupting oxidative photosynthesis processes, decreases in stomatal conductance, reduction in chlorophyll content, reduced carboxylation efficiency, energy budgets of the plant, decrease in Rubisco activity, affecting soluble protein, reduces growth and yield response Many metabolic pathways are altered by O<sub>3</sub>. Ozone can cause different effects on different protein pools (Friend et al., 1992). Studies of Rubisco chemistry from O<sub>3</sub> treated leaves suggest

oxidative followed by rapid protease digestion (Junqua et al., 2000). Functional and structural breakdown of chloroplast leads photo inhibition, loss of ribulose and rubisco protein (Heath et al., 1997) in the apoplast and in the outer surface of the plasmalemma damaged through O<sub>3</sub> exposure (Mudd et al, 1997a). Three amino acid residues are particularly sensitive to ozone. Ozonolysis will open up the pyrrol ring of tryptophan and oxidize the sulphhydryl group (- SH) of cysteine and methionine to form disulphide bridges (-S-S) Sulphoxides (Wellburn, 1994; Mudd et al., 1997b). Similarly in the present study decrease in protein content in Tomato plant was recorded. The loss in Tomato plant was 16.45 %. (Table-1).

#### Plant Age 62 Days

After considerable reduction of protein content under the O<sub>3</sub> of gaseous exposure, significant recovery observed on this plant age which suggests that Sodium Erythroate antidote treatment mitigate the adverse effect of air pollution. Plant antioxidant system, which scavenges naturally, occurring ROS (Reactive Oxygen Species) compounds, could function as primary mechanism to

**Table 1 : Effect of O<sub>3</sub> (60µg/m<sup>3</sup>) on Foliar Protein (mg/g Fresh wt.) in Tomato Plant and Their Amelioration by N, N'-Dimethylurea (DMU) Antidote**

Plant age (In days)	Dose	N, N'-Dimethylurea (DMU)Antidote (ppm)
30 days	Control	06.14± 0.15
51 days Gaseous treatment for 20 days	Control	09.36± 0.65
	Treatment	07.82± 0.48
Loss <sub>.1</sub> at 1 <sup>st</sup> stage (%)		<b>16.45</b>
62 days Antidote treatment and 10 days for recovery	Control	13.81 ± 0.12
	20 ppm	18.44 ±0.73
	100 ppm	19.85 ±1.41
Recovery <sub>.1</sub> at 1 <sup>st</sup> stage (%)	20 ppm	<b>49.24</b>
	100 ppm	<b>52.84</b>
83 days Gaseous treatment for 20 days	Control	14.31 ±0.08
	20 ppm	10.33 ±0.17
	100 ppm	13.45±0.56
Loss <sub>.2</sub> at 2 <sup>nd</sup> stage (%)	20 ppm	<b>43.98</b>
	100 ppm	<b>33.24</b>
94 days Antidote treatment and 10 days for recovery	Control	17.83 ±3.87
	20 ppm	12.79 ±0.14
	100 ppm	15.33 ±0.61
Recovery <sub>.2</sub> at 2 <sup>nd</sup> stage (%)	20 ppm	<b>19.23</b>
	100 ppm	<b>12.26</b>

±: Standard Deviations  
DMU - N, N'-Dimethylurea

alleviate the oxidative burden resulting from ozone exposure. Reactive oxygen species may also act as a signal to initiate or coordinate other processes such as ethylene production, which induces senescence (Pell et al., 1997; Schraudner et al., 1997). Some workers reported that applications of two recently developed fungicides (azoxystrobin and epoxiconazole) provided protection to spring barley to relatively high O<sub>3</sub> exposures and resulted in increases in leaf soluble protein content as well as the activity of several antioxidative enzymes (e.g., Superoxide dismutase, catalase, ascorbate-peroxidase, and glutathione reductase).

N, N'-Dimethylurea (DMU) used during the study was found effective and could reduce the deleterious effect of ozone in Tomato crops. The recovery observed after loss in Tomato plant low dose and high dose of DMU brings recovery of 49.24% and 52.84% respectively. (Table-1).

#### **Plant Age 83 Days**

Plant responses depend not only on the inherent characters of plant species but also on the stage of development, age and nutritional status (Kovacs, 1992). On this age O<sub>3</sub> shows considerable loss in the protein content in Tomato crops. Tropospheric ozone has profound negative impacts on the growth, development and productivity of plants. Many metabolic pathways are also altered by ozone exposure. A toxic product of O<sub>3</sub> may migrate through the cytoplasm to react with photosynthetic processes, or a spurious signal generated at the membrane may affect some control process or signal transduction pathway, which ultimately alters the carbon assimilation and protein synthesis. (Schraudner et al., 1998; Overmyer et al., 2000, 2003; De Caria et al., 2000; Rao et al., 2002; Booker et al., 2004; Leitao et al., 2003; Rao and Davis, 2001; Sandermann, 2000; Vahala et al., 2003). The maximum loss in protein in Tomato plant at 20-ppm was 43.98 % and 100-ppm was 33.24%. (Table-1)

#### **Plant Age 92 Days**

Significant recovery in protein content after antidote treatment in O<sub>3</sub> of gaseous exposure was observed which suggest that antidotes treatment must be promising in ameliorating the effect of air pollutants and allows the plants grow under natural conditions. With the application of

different antidotes significant recovery was observed in the studied crops. In Tomato plant at 20- ppm the minimum recovery observed was 19.23% and 100-ppm the minimum recovery observed was 12.26 %. (Table-1) The use of antidotes and fertilizer may inhibit response of O<sub>3</sub> and protects crops from the damage (Haryani, K. 2002; Dubey, 2002).

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